CHLORINE

CAUTIONARY RESPONSE INFORMATION Common Synonyms Liquefied compressed Greenish yellow Irritating, bleachlike choking odor Sinks and boils in water. Poisonous, visible vapor cloud is produced. Keep people away. AVOID CONTACT WITH LIQUID AND VAPOR. Wear goggles, self-contained breathing apparatus and rubber overclothing (including gloves). Evacuate area in case of large discharge. Stay upwind and use water spray to "knock down" vapor. Notify local health and pollution control agencies. Protect water intakes. Not flammable Fire May cause fire on contact with combustibles. POISONOUS GASES ARE PRODUCED IN FIRES. Stop flow of gas if possible. Cool exposed containers and protect men effecting shutoff with water. Wear goggles, self-contained breathing apparatus, and rubber overclothing (including gloves). CALL FOR MEDICAL AID. **Exposure** VAPOR POISONOUS IF INHALED. Will burn eyes. Move to fresh air. If breathing has stopped, give artificial respiration (but NOT mouth-to-mouth). If the atthing is difficult, give oxygen. IF IN EYES, hold eyelids open and flush with plenty of water. LIQUID Will burn skin and eves. Will cause frostbite. Flush affected areas with plenty of water. FI IN FYES, hold eyelids open and flush with plenty of water. DO NOT RUB AFFECTED AREAS. HARMFUL TO AQUATIC LIFE IN VERY LOW CONCENTRATIONS. Water May be dangerous if it enters water intake Notify local health and wildlife officials. **Pollution**

1.	CORRECTIVE	RESPONSE	ACTIONS

Dilute and disperse Stop discharge

Do not add water to undissolved material

2. CHEMICAL DESIGNATIONS

- CG Compatibility Group: Not listed.
- 2.3

- CG Compatibility Group: Not insied.
 Formula: Cla
 IMO/UN Designation: 2.0/1017
 DOT ID No.: 1017
 CAS Registry No.: 7782-50-5
 NAERG Guide No.: 124
 Standard Industrial Trade Classification:
 52274 52224

3. HEALTH HAZARDS

Notify operators of nearby water intakes

- 3.1 Personal Protective Equipment: Quick-opening safety shower and eye fountain; respiratory equipment approved for chlorine service. Wear safety goggles at all times when in vicinity of liquid chlorine.
- 3.2 Symptoms Following Exposure: Eye irritation, sneezing, copious salivation, general excitement and restlessness. Irritation may persist for several days. High concentrations cause respiratory distress and violent coughing, often with retching. Death may result from suffocation.
- 3.3 Treatment of Exposure: INHALATION: remove victim from source of exposure; call a doctor; support respiration; administer oxygen. EYES: flush with copious amounts of water for at least 15 min.
 3.4 TLV-TWA: 0.5 ppm
- 3.5 TLV-STEL: Not listed.
- 3.6 TLV-Ceiling: 1 ppm.
- 3.7 Toxicity by Ingestion: Not pertinent; ingestion unlikely (chlorine is a gas above -34.5°C)
- 3.8 Toxicity by Inhalation: Currently not available.
- 3.9 Chronic Toxicity: None
- 3.10 Vapor (Gas) Irritant Characteristics: Vapors cause severe irritation of eves and throat and can cause eye and lung injury. They cannot be tolerated even at low concentrations.

 3.11 Liquid or Solid Characteristics: Causes smarting of the skin and first-degree burns on short
- exposure; may cause secondary burns on long exposure.

 3.12 Odor Threshold: 3.5 ppm
- 3.13 IDLH Value: 10 ppm
- 3.14 OSHA PEL-TWA: Not listed.
- 3 15 OSHA PEL-STEL: Not listed
- 3.16 OSHA PEL-Ceiling: 1 ppm.
- 3.17 EPA AEGL: Not listed

4. FIRE HAZARDS

- 4.1 Flash Point: Not flammable
- 4.2 Flammable Limits in Air: Not flammable
- 4.3 Fire Extinguishing Agents: Currently not
- 4.4 Fire Extinguishing Agents Not to Be Used: Not pertinent
- 4.5 Special Hazards of Combustion Products: Toxic products are generated when combustibles burn in chlorine.
- **Behavior in Fire:** Most combustibles will burn in chlorine, although gas is not flammable.
- 4.7 Auto Ignition Temperature: Not
- 4.8 Electrical Hazards: Currently not
- 4.9 Burning Rate: Not flammable
- 4.10 Adiabatic Flame Temperature: Currently not available
- 4.11 Stoichometric Air to Fuel Ratio: Not Pertinent
- 4.12 Flame Temperature: Currently not available
- 4.13 Combustion Molar Ratio (Reactant to Product): Not Pertinent
- Minimum Oxygen Concentration for Combustion (MOCC): Not listed

5. CHEMICAL REACTIVITY

- Reactivity with Water: Forms a corrosive solution

- 5.5 Polymerization: Not pertinent

6. WATER POLLUTION

- 6.2 Waterfowl Toxicity: Currently not

- 6.5 GESAMP Hazard Profile:

7. SHIPPING INFORMATION

- 7.1 Grades of Purity: Research purity; ultra high purity: high purity
- 7.2 Storage Temperature: Ambient
- 7.3 Inert Atmosphere: No requirement 7.4 Venting: Safety relief (300 psi)
- 7.5 IMO Pollution Category: Currently not available
- 7.6 Ship Type: Currently not available
- 7.7 Barge Hull Type: 1

8. HAZARD CLASSIFICATIONS

- 8.1 49 CFR Category: Poison Gas
- 8 2 49 CFR Class: 2 3
- 8.3 49 CFR Package Group: Not pertinent
- 8.4 Marine Pollutant: Yes
- 8.5 NFPA Hazard Classification:

Category			
Health Hazard (Blue)) 3		
Flammability (Red)	0		
Instability (Yellow)	0		
Special (White)	0	Х	

- 8.6 EPA Reportable Quantity: Not listed.
- 8.7 EPA Pollution Category: Not listed.
- 8.8 RCRA Waste Number: Not listed
- 8.9 EPA FWPCA List: Not listed

- 5.2 Reactivity with Common Materials: Reacts vigorously with most metals at high temperature. Copper may burn spontaneously.
- 5.3 Stability During Transport: Stable
- 5.4 Neutralizing Agents for Acids and Caustics: Not pertinent
- 5.6 Inhibitor of Polymerization: Not pertinent

- 6.1 Aquatic Toxicity:
 0.08 ppm/168 hr/trout/TLm/fresh water
 10 ppm/1 hr/tunicates/killed/salt water
- 6.3 Biological Oxygen Demand (BOD): None
- 6.4 Food Chain Concentration Potential:
- Bioaccumulation: 0
 Damage to living resources: 4
 Human Oral hazard: N/A
 Human Contact hazard: II Reduction of amenities: XX

9. PHYSICAL & CHEMICAL **PROPERTIES**

- 9.1 Physical State at 15° C and 1 atm: Gas
- 9.2 Molecular Weight: 70.91
- 9.3 Boiling Point at 1 atm: -29.4°F = -34.1°C = 239.1°K
- 9.4 Freezing Point: -150°F = -101°C = 172°K
- 9.5 Critical Temperature: 291.2°F = 144°C = 417.2°K
- **9.6 Critical Pressure:** 1118 psia = 76.05 atm = 7.704 MN/m²
- 9.7 Specific Gravity: 1.424 at 15°C (liquid)
- 9.8 Liquid Surface Tension: 26.55 dynes/cm at -35.3°C
- 9.9 Liquid Water Interfacial Tension: Not pertinent
- 9.10 Vapor (Gas) Specific Gravity: 2.4 9.11 Ratio of Specific Heats of Vapor (Gas):
- 1.325
- 9.12 Latent Heat of Vaporization: 124 Btu/lb =
- 68.7 cal/g = 2.87 X 10⁵ J/kg 9.13 Heat of Combustion: Not pertinent
- 9.14 Heat of Decomposition: Not pertinent 9.15 Heat of Solution: Not pertinent
- 9.16 Heat of Polymerization: Not pertinent
- 9.17 Heat of Fusion: 22.8 cal/g
- 9.18 Limiting Value: Currently not available 9.19 Reid Vapor Pressure: 155 psia

NOTES

CHLORINE

9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
-35 -30	97.179 96.730		N O T		N O T		N O T
			P E R T I N E N T		P E R T I N E N T		P E R T - N E N T

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
77	0.650	-55 -50 -45 -40 -35 -30 -25 -20 -15 -10 -5 0 5 10 15 20 22 30 35 40 45 50	7.491 8.580 9.795 11.150 12.650 14.310 16.140 18.150 20.370 22.790 25.450 28.340 31.490 34.910 38.610 42.620 46.950 51.620 56.650 62.050 67.839 74.040	-55 -50 -45 -40 -35 -30 -25 -20 -15 -10 -5 0 5 10 15 20 22 5 30 35 40 45 50	0.12230 0.13840 0.15600 0.17550 0.19670 0.22000 0.24530 0.27280 0.30260 0.33490 0.36970 0.40720 0.44760 0.49090 0.53730 0.63990 0.63990 0.69640 0.75650 0.82030 0.88790 0.95960	0 25 50 75 100 125 125 125 125 125 125 125 125 125 125	0.112 0.113 0.114 0.114 0.114 0.115 0.116 0.116 0.117 0.118 0.119 0.120 0.120 0.120 0.121 0.121 0.122 0.122 0.122 0.122 0.122 0.122 0.122 0.122 0.122